

The Activation of Phosphoprotein Phosphatase with Inorganic Phosphate and Surfactants

BO NORBERG

Chemical central laboratory, Sabbatsbergs sjukhus, Stockholm, Sweden

In the determination of the activity of phosphoprotein phosphatase as released inorganic phosphate, Pi, which can be precipitated with calcium at pH 9, the enzyme activity is dependent upon the concentration of Pi, so that the apparent activity approaches 0 in connection with falling Pi, attains a maximum at about 0.1 mM Pi, when the substrate is 1 % phosvitin, and sinks in connection with an even moderate excess of Pi. The enzyme activity is further increased by quaternary ammonium bases such as cetyltrimethylammonium bromide and other detergents. The activation is presumed to be connected with an increased tendency to reaction in the carbinol-group of the serine residue in the substrate.

In an earlier investigation¹ it was found that precipitation of the inorganic phosphate (Pi) released by the phosphoprotein phosphatase was necessary in order to get satisfactory reproducibility and proportionality between enzyme concentration and reaction rate. In the course of subsequent work, however, a number of apparently inexplicable irregularities were observed. Thus on one day fresh liver extract gave good activity with two casein substrates (2.17 and 3.86 units per gram liver), whereas much smaller values were obtained the next day (0 and 1.17 units per gram). Experiments with phosvitin² also gave unexpected results, adrenal extract giving higher activity, in 5 out of 6 experiments, than liver extract, as compared with the reverse relation when casein was used as substrate¹. Parallel measurements of activity were then carried out with casein and phosvitin as substrate, and extract of adrenal glands, erythrocytes and liver from the rat, guinea-pig, rabbit, and cat as enzyme source. With liver in these experiments a higher apparent activity was obtained with casein substrate in 11 out of 13 tests. Adrenals and blood corpuscles, on the other hand, gave higher activity with phosvitin in 10 out of 15 experiments. Since inhibition of phosphoprotein phosphatase with Pi has been shown^{1,3}, and since liver is richer in Pi than are red corpuscles, the irregularities might possibly be due to the occurrence of Pi. Attempts to eliminate the excess Pi by dialysis, however, resulted regularly in a loss of activity. A more careful investigation of the effect of Pi was therefore indicated.

In the subsequent experiments human erythrocytes were chiefly used as source of enzyme. In a preliminary series with rat liver, an extract in 0.1 M acetate, of pH 5.8, gave activities of between 3.26 and 4.76 units per gram, on an average 4.07 with a standard deviation of 0.056, while an aqueous extract gave 3.88–4.87 units per gram, averaging 4.35. Systematic testing with erythrocyte extract showed optimal activity at an ionic strength of 0.04. Accordingly the erythrocytes, after 3 washings with isotonic sodium chloride were hemolyzed with distilled water, after which the hemolysate was diluted to 50 mg hemoglobin per ml and ionic strength 0.2, using sodium chloride and acetate buffer, pH 5.8. Phosvitin with about 1.1 mg protein nitrogen per ml (determined on trichloroacetic acid precipitate) was used as substrate.

Measurement of the enzyme activity was performed by incubating 1 ml extract and 4 ml substrate at 38°C, then stopping the enzyme effect with trichloroacetic acid and measuring the Pi released after precipitation with calcium at pH 9¹. Apart from this standard method two alternatives were tried: a) direct determination of Pi in the trichloroacetic acid extract according to Fiske-Subbarow and b) extraction of directly obtained phosphomolybdate complex⁴. However, these alternative methods regularly gave lower values than did the standard method, especially at the optimal initial concentration of Pi. In 84 comparisons the standard technique gave on an average 1.61 ± 0.27 (standard deviation) times greater activity than did alternative a). The apparent difference in activity is referable to the higher blank value in the alternative methods. The preliminary explanation of this might be the occurrence of an acid-labile phosphorus linkage in the substrate. In subsequent testing only the standard technique was used.

The recovery of Pi which had been added to dialysed substrate was investigated at four phosvitin concentrations. It was found that in order to get a certain blank value more Pi was required the higher the phosvitin-content. This emerges clearly from Fig. 1, where increasing addition of phosphate does not give a rising blank value until after the addition of 100 μ mole/l with approx. 1 % phosvitin, and of 200 μ mole/l for 2 % phosvitin.

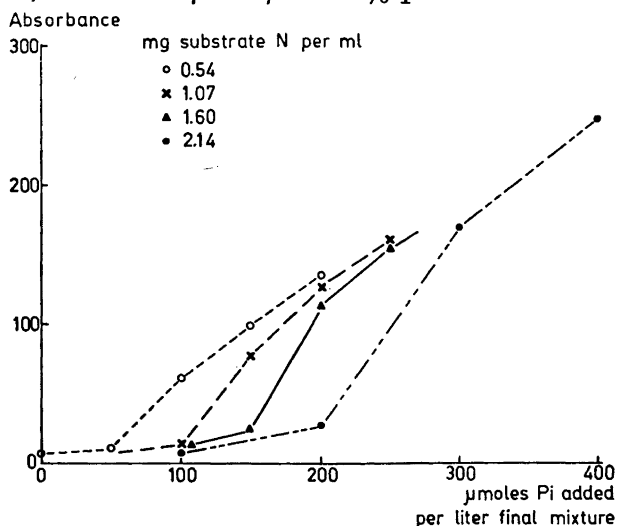


Fig. 1. Apparent inorganic phosphate (Pi, as absorbance) against total concentration of added Pi at increasing phosvitin concentrations.

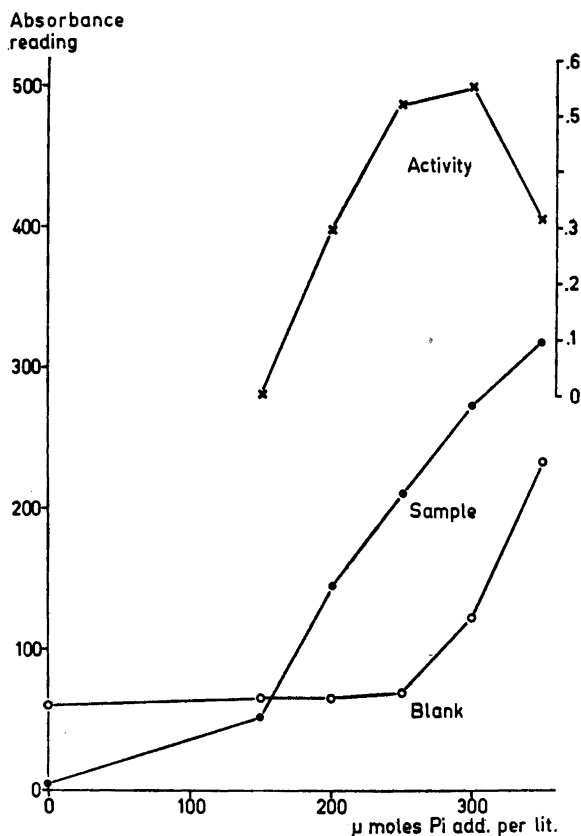


Fig. 2. Effect of addition of inorganic phosphate on blank reading and enzyme activity.

The relation between enzyme activity and added Pi is illustrated by the experiment shown in Fig. 2. On the addition of Pi corresponding to up to 0.15 mM no enzyme activity at all is obtained on incubation. On larger additions it increases however, reaching a maximum at the same time as the blank value begins to rise in response to a Pi-addition corresponding to a final concentration of 0.25 mM phosphate at 1.8 % phosvitin. Even slightly larger amounts of Pi result in inhibition.

In subsequent experiments 1 % phosvitin was usually used as substrate, and maximal enzyme activity was then obtained at a blank value of about 100 μ M Pi. This appears from a series of experiments comprising 39 tests and reproduced in Fig. 3. When the actual concentration of Pi found in the blank value exceeds 150 μ M, the activity is greatly reduced. If the initial Pi-value approaches 0, the enzyme activity disappears entirely. One must obviously determine the content of Pi in every enzyme sample and adapt the addition of Pi to the substrate so that the optimal concentration of Pi for the actual substrate is approximately reached; the interval between too little and too much is evidently very small.

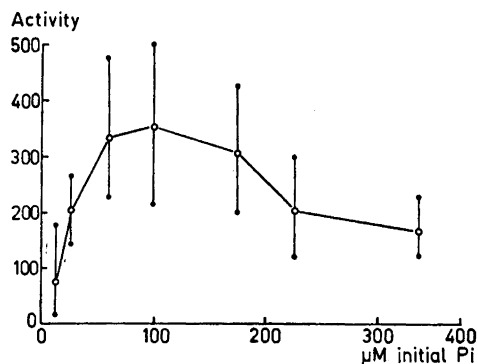


Fig. 3. Actual initial concentration of Pi at incubation and its effect on enzyme activity (mean value and range). Each mean represents 2 to 8 experiments.

Just how sensitive the enzyme reaction is to excess phosphate is shown in Fig. 4. With a rising incubation period an increasing concentration of Pi is obtained, which results in increasing inhibition; so that the apparent activity per 100 mg hemoglobin has fallen below 50 % of the highest value measured when the concentration of Pi has risen to no more than 0.4 mM. This striking sensitivity to phosphate in the incubation milieu shows clearly how difficult it is with *in vitro* experiments to throw light upon the activity of the enzyme *in vivo*.

In connection with the testing of casein as substrate, *quaternary ammonium bases* were also added in the dialysis, with the intention of preventing bacterial contamination of the substrate. This resulted, in the majority of the experiments in a not inconsiderable increase in the activity, as is exemplified in Table 1.

Similar results were also obtained with octylphenoxyethoxyethyl-dimethylbenzylammonium chloride ("Septin" Pharmacia, Uppsala) and cetyltrimethylammonium bromide, CTB, and also with ethylenediamine-tetraacetate, EDTA. The sensitivity of the enzyme to the phosphate ion concentration was not investigated however, and with the exception of some of these tests no attention was paid to this factor. The significance of the addition of phosphate for casein which had been dialyzed against 0.5 % CPBr with extract from fresh rat's liver as enzyme is shown in Table 2. In other casein substrates which had been dialyzed with or without addition of quaternary ammonium base an addition of phosphate corresponding to 0.12–1.05 mM Pi was required at the start of the incubation, with an average of 0.51 mM Pi for optimal effect (22 substrates with 20 mM organically bound phosphorus).

For testing with phosvitin CTB in a concentration of 10 mM was generally used. The effect of the detergent and of the addition of phosphate is shown in Table 3, which clearly indicates stimulation but also the dependence on the addition of phosphate both without and with CTB. To throw further light upon this, Table 4 shows the results of an experiment in which the phosvitin solutions were prepared directly in water and in 0.01 M CTB, respectively, without dialysis. Here the water substrate gives an apparently very low

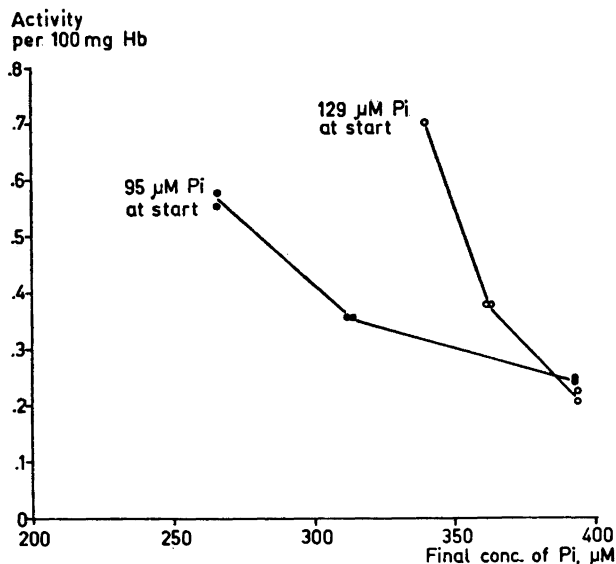


Fig. 4. Increasing inhibition of enzyme activity with rising concentration of Pi in digest. Initial concentration of Pi 95 μM (dots) and 129 μM (circles).

concentration of Pi. In the CTB-substrate, on the other hand, one obtains with increasing phosphitin-concentration a considerable Pi concentration. The optimal activity is obtained at an apparent concentration of 61 μM Pi in the substrate + 214 μM in the hemolysate. The activation, which apparently derives from the detergent, seems to be intimately connected with the Pi which is available, and at a higher substrate-concentration it turns into a faint but distinct inhibition. The interesting thing about this experiment is that the detergent appears to mobilize the labile phosphorus in the substrate, as there was no special addition of phosphate.

The results of the investigation appear to show that a certain minimal concentration of inorganic phosphorus is required to obtain the greatest possible activity with phosphoprotein phosphatase in the measuring system used. On the other hand, further inorganic phosphorus which is produced by the

Table 1. Phosphoprotein phosphatase activity of rat liver homogenate using casein dialyzed against acetate buffer with or without cetylpyridinium bromide, CPBr.

Casein No.	CPBr concentration at dialysis, mM			Maximum change, %
	None	0.26–1.3	2.6–13	
13	0.74	0.59	0.61	– 20
14	1.70	2.06	2.19	+ 29
17	0.52	3.94	4.82	+ 825
19	0	1.20	1.01	+ ∞
21	5.13	4.64	—	– 10
33	1.17	1.86	2.59	+ 120
34	10.5	11.6	—	+ 10
43	9.72	14.5	—	+ 49

Table 2. Effect of addition of inorganic phosphate, Pi, to casein dialyzed against 0.5 % CPBr on activity of liver phosphoprotein phosphatase.

Pi added corresponding to concentration at incubation, μM	Activity in units per gram of fresh liver
0	0
100	0.1
200	0.02
300	5.01
400	7.74
500	9.40
600	10.7

enzyme or which exists from the outset, results in a striking inhibition. It is evident that this phosphatase activity can be regulated with great sensitivity by Pi. A further activation through detergents also seems to occur, and the experiments appear to indicate that this effect may consist in a loosening and breaking of covalent phosphorus linkages in the substrate.

The *in vitro* system applied cannot be considered comparable with the conditions obtaining *in vivo*, but it is an interesting coincidence that at the maximal activation of the enzyme the concentration of Pi coincides with the minimal concentration required for oxidative phosphorylation in respiring mitochondria⁵⁻⁷. In contradistinction to oxidative phosphorylation and phosphorylation⁸, where Pi is included as substrate component, the activating phosphate seems rather to act as a cofactor for the phosphoprotein phosphatase, as is the case in the transformation of serine to glycine⁹. The possibility that Pi is required to overcome a solubilization effect in the precipitation of calcium phosphate¹⁰, which is included in the analytic procedure applied, cannot be ruled out, but it seems unlikely, since the phosphate effect exists, even if more faintly, in direct phosphate analysis. On the other hand, it is not inconceivable that through dialysis the substrate loses some phosphate, which is replaced

Table 3. Effect of added KH_2PO_4 on activity of erythrocyte phosphoprotein phosphatase on phosvitin prepared in water and in 0.01 M cetyltrimethylammonium bromide, CTB.

Added Pi corresponding to final conc., μM	Activity per 100 mg hemoglobin		% increase with CTB
	1.8 % phosvitin in water	2.3 % phosvitin in 0.01 M CTB	
0	0	0	0
150	0	0.123	∞
200	0.292	0.490	68
250	0.521	0.732	40
300	0.554	0.883	60
350	0.309	0.575	86

Table 4. Comparison between activities of a hemolysate on phosphovitin of different concentrations without and with 0.01 M CTB.

Substrate % phosphovitin	Substrate in water		Substrate in 0.01 M CTB		% change of activity with CTB
	Apparent Pi in substrate, μ M	Activity U/100 mg hemoglobin	Apparent Pi in substrate, μ M	Activity U/100 mg hemoglobin	
1	7	0.128	15	0.272	+106
2	9	0.194	61	0.334	+ 72
3	13	0.199	268	0.295	+ 48
4	14	0.195	470	0.184	- 5
5	15	0.241	524	0.178	- 26
hemolysate	214				

by added Pi. A simple and rapid exchange between serine-bound phosphorus and Pi has been shown¹¹, and, as has been adduced above, it is conceivable that the detergent effect is connected with a further loosening of the phosphorus-serine linkage or with activation of the carbinol-group of the serine¹². In view of the fact that surfactants prove as a rule to be enzyme-inhibitors¹³ it is also more probable that the effect here starts on the substrate; this would entail an intensification of the affinity between enzyme and substrate¹⁴. The inhibition of excess Pi may be a competition for the active site of the enzyme.

Acknowledgements. This investigation has been supported by the *Swedish Medical Research Council*, and by *The National Institutes of Health of the United States* (Grant H-1142) during studies at the Department of Surgical Research, Medical College of Virginia, Richmond, Va. The author is indebted to Miss Edith Andersson for devoted assistance.

REFERENCES

1. Norberg, B. *Acta Chem. Scand.* **4** (1950) 1206.
2. Mecham, D. K. and Olcott, H. S. *J. Am. Chem. Soc.* **71** (1949) 3670.
3. Singer, M. F. and Fruton, J. S. *J. Biol. Chem.* **299** (1957) 111.
4. Ernster, L., Zetterström, R. and Lindberg, O. *Acta Chem. Scand.* **4** (1952) 804.
5. Lardy, H. A. and Wellman, H. *J. Biol. Chem.* **195** (1952) 215.
6. Gatt, S. and Racker, E. *J. Biol. Chem.* **234** (1959) 1015, 1024.
7. Wadkins, C. L. and Lehninger, A. L. *J. Biol. Chem.* **234** (1959) 681.
8. Rowen, J. W. and Kornberg, A. *J. Biol. Chem.* **193** (1951) 497.
9. Wright, B. E. and Stadtman, T. C. *J. Biol. Chem.* **219** (1956) 863.
10. Mandl, I., Grauer, A. and Neuberger, C. *Biochim. et Biophys. Acta* **10** (1953) 540.
11. Ågren, G. *Acta Chem. Scand.* **13** (1959) 1048.
12. Koshland, D. E. and Erwin, M. Y. *J. Am. Chem. Soc.* **79** (1957) 2659.
13. Wills, E. D. *Biochem. J.* **60** (1955) 529.
14. Hughes, D. E. *Biochem. J.* **45** (1949) 325.

Received November 5, 1959.